



SRI CHAITANYA EDUCATIONAL INSTITUTIONS,INDIA.

A.P,TELANGANA,KARNATAKA,TAMILNADU,MAHARASHTRA,DELHI,RANCHI,CHANDIGARH

SEC : SR. OUTGOING ALL STREAMS

DATE : 28.08.2020

SUB : PHYSICS

NEET GRAND TEST-19

91. Read the following statements and Identify the correct statement/s.

A: A choice of change of different units change the number of significant digits.

B: To remove ambiguities in determining the number of significant figures, the best way is to report the measurement in scientific notation.

C: Solid angle made by a hemisphere at its centre is 4π steradian.

None of the above is true

None of the above is true

Both B and C are true

Both B and C are true

Key : 2

Solution : Change of units never change significant figures.

$$\text{Solid angle} = \frac{\text{Surface area}}{r^2} = \frac{4\pi r^2}{r^2} = 4\pi$$

92. A particle moving on a straight line velocity is given by $V = e^{8t}$ then acceleration of particle when velocity is 10 m/s is

$8 \times 10 \text{ m/s}^2$

$8 \times 10^8 \text{ m/s}^2$

$8 \times 10 \text{ m/s}^2$

$8 \times 10^8 \text{ m/s}^2$

Key : 1

Solution : $V = e^{8t}$

$$\frac{dv}{dt} = 8e^{8t}$$

$$a = \frac{dv}{dt} = 8 \frac{dv}{V}$$

$$a = 8 \times V \rightarrow (1)$$

Given $V = 10$.

$$8 \times 10 = 80 \Rightarrow e^{8t} = 10$$

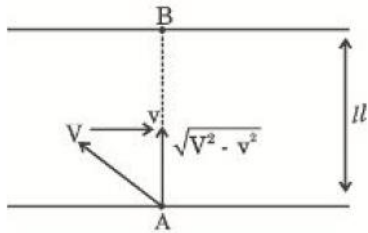
$$(1) \Rightarrow a = 10 \times 8 \times 10 = 800 \text{ m/s}^2$$

93. An aeroplane is to go along straight line from A to B, and back again. The relative speed with respect to wind is V . The wind blows perpendicular to line AB with speed v . The distance between A and B is l . The total time for the round trip is

$\frac{l}{\sqrt{V^2 - v^2}}$
 $\frac{vl}{V^2 - v^2}$
 $\frac{Vl}{V^2 - v^2}$
 $\frac{l}{\sqrt{V^2 + v^2}}$

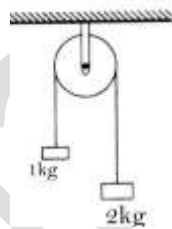
Key : 1

Solution :



$$\text{time} = \frac{l}{\sqrt{V^2 - v^2}} + \frac{l}{\sqrt{V^2 - v^2}} = \frac{2l}{\sqrt{V^2 - v^2}}$$

94. Two unequal masses 1kg and 2 kg are connected on two sides of light string passing over a light and smooth pulley as shown. The system is released from rest. The larger mass is stopped for a moment, 1.0 sec after the system is in motion. The time elapsed before the string is tight again is (in sec)



$\frac{1}{3}$
 $\frac{2}{3}$
 $\frac{1}{2}$
 $\frac{2}{5}$

Key : 4

Solution : attwoods machine, $a = \left(\frac{2-1}{2+1}\right)g = \frac{g}{3}$ $V = u + at = 0 + (g/3)1 = g/3$

For 1 kg mass: $h_1 = vt - \frac{1}{2}gt^2 = \left(\frac{g}{3}\right)t - \frac{1}{2}gt^2$

For 2 kg mass: $h_2 = \frac{1}{2}gt^2$ If $h_2 = h_1$ then $t = 1/3 \text{ sec}$

95. A ball is thrown with a velocity of 6m/s vertically downwards from a height $H=3.2\text{m}$ above a horizontal floor. If it rebounds back to same height then coefficient of restitution e is $\frac{1}{3}g = \frac{1}{3}m \frac{1}{s^2}$

$\frac{1}{3}$
 $\frac{2}{3}$
 $\frac{1}{2}$
 $\frac{2}{5}$

Key : 4

Solution : $V = 10$ e = $\frac{V_1}{10}$

$$V_1 = 10e$$

$$v_1^2 = 4^2 + 2as$$

$$100e^2 = 2 \times 10 \times 3.2$$

$$e = 0.8$$

96. The force acting on the block is given by $F = (5 - 2t) \text{ N}$. The frictional force acting on the block at time $t = 2 \text{ s}$ will be $\mu \times$



10 N

2 N

1 N

5 N

Key : 1

Solution : At $t = 0$, $F = 5 \text{ N}$

$$f = 0.2 \times 1 \times 10 = 2 \text{ N}$$

As $F > f$, it in motion friction is kinetic

At $t = 2 \text{ sec}$, $F = 1 \text{ N}$,

As $F < f$, body will move with retardation, but still friction is kinetic i.e., 2N only.

Kinetic friction is not self adjusting.

97. When a ceiling fan is switched on, it makes 10 rotations in the first 3 seconds. How many rotations will it make in the next 3 seconds? (Assume uniform angular acceleration.)

10

20

30

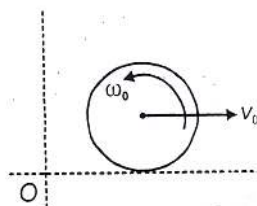
40

Key : D

$$\omega^2 - \omega_0^2 = \alpha \propto \theta$$

Solution :

98. A hollow sphere having mass m and radius R is rolling as shown in the figure. If the speed of centre of mass of sphere is V_0 and angular speed is $\omega_0 = \frac{V_0}{R}$. The angular momentum of the sphere about point O is



$\frac{1}{2} MV_0 R$

$MV_0 R$

$\frac{MV_0 R}{2}$

$\frac{MV_0 R}{3}$

Key : 4

Solution : $L_o = \frac{2}{3}MR^2 \times \frac{V_o}{R} \sim MV_oR = \frac{MV_oR}{3}$

99. A wheel of radius R rolls on the ground with a uniform velocity 'v'. The relative acceleration of topmost point of the wheel with respect to the bottom most point is

$\frac{v^2}{R}$

$\frac{v^2}{R}$

$\frac{v^2}{R}$

$\frac{v^2}{R}$

Key : 2

Solution : Rel. acc of top most point w.r.t. bottom most

$$a = \frac{(\text{---})^2}{\text{---}} = \frac{\text{---}^2}{\text{---}}$$

100. A body of mass 'M' is moving on a circular track of radius 'r' in such a way that its kinetic energy 'k' depends on the distance travelled by the body 's' according to relation $k = \beta s$, where β is a constant. The angular acceleration of the particle is

$\frac{\beta r}{M^2}$

$\sqrt{\frac{\beta r}{M}}$

$\frac{M^2 r}{\beta}$

$\frac{\beta}{Mr}$

Key : 4

Solution : $K = \beta s \quad \frac{1}{2}mv^2 = \beta s$

$$a_t = \frac{\beta}{M} \Rightarrow r\alpha = \frac{\beta}{M}$$

$$\alpha = \frac{\beta}{Mr}$$

101. A particle undergoes SHM with a time period of 2 seconds. In how much time will it travel from its mean position to a displacement equal to half of its amplitude?

$\frac{1}{3} \text{ s}$

$\frac{1}{3} \text{ s}$

$\frac{1}{3} \text{ s}$

$\frac{1}{3} \text{ s}$

Key : D

Solution : $y = A \sin \omega t$

Here $y = \frac{A}{2}$

102. Two waves travelling in a medium in the x-direction are represented by $y_1 = a \sin(\alpha t - \beta x)$ and $y_2 = a \sin(\beta x + \alpha t - \pi)$, where y_1 and y_2 are the displacements of the particles of the medium, t is time, and α and β are constants.

The two waves have different

(A) Speeds

(B) Directions of propagation

(C) Wavelengths

(D) Periods

Key : B

Solution : The first wave is propagating along positive X-axis, the second is along negative X – axis. But both the waves have same speed, frequency and wavelength

103. An astronaut of mass m is working in a satellite orbiting the earth at a distance h from the earth's surface. The radius of the earth is R, while its mass is M. The gravitational pull F_G on the astronaut is

(A) Zero (B) $\frac{GMm}{(R+h)^2}$ (C) $\frac{GMm}{R^2}$ (D) $\frac{GMm}{R^2} < F_G < \frac{GMm}{(R+h)^2}$

$$F_G = \frac{GMm}{(R+h)^2}$$

$$\frac{GMm}{R^2} < F_G < \frac{GMm}{(R+h)^2}$$

Key : 3

$$F_g = \frac{M_1 M_2}{d^2}$$

(A) A car is moving with a uniform speed on a level road. Inside the car there is a balloon filled with helium and attached to a piece of string tied to the floor. The string is observed to be vertical. The car now takes a left turn maintaining the speed on the level road. The balloon in the car will (Here the car is supposed to be air tight)

(A) Continue to remain vertical

(B) First tilt to the right

(C) Tilt to the right side

(D) Tilt to the left side

Key : 4

Solution : Due to pseudo force.

105. A raindrop reaching the ground with terminal velocity has momentum p. Another drop of twice the radius, also reaching the ground with terminal velocity, will have momentum

(A) p

(B) 2p

(C) 8p

(D) 16p

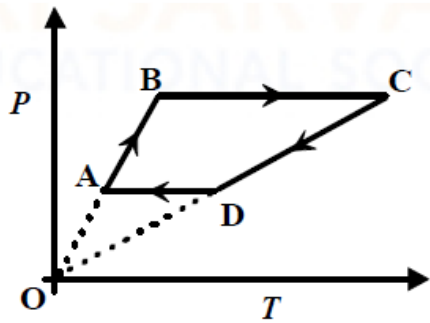
Key : D

Solution : $P = mv$

$$v \propto r^2$$

$$P \propto r^3$$

106. One mole of an ideal monatomic gas performs a cyclic process indicated by ABCDA. The temperatures of the gas at A, B, C and D are respectively T, 2T, 6T and 3T. If R is the molar gas constant, the work done by the gas in the cyclic process is



- 2RT
 3RT
 4RT
 5RT

Key : 3

Solution :
 $\omega_{AB} = 0$
 $\omega_{BC} = p \Delta V = nR \Delta T$
 $= R P.(4T)$
 $\omega_{CD} = 0$
 $\omega_{DA} = p \Delta V = nR \Delta T$
 $= R P.(T-3T)$
 $= -2RT$
Total
 $\omega = (4RT - 2RT)$
 $= 2RT$

107. In a p-n junction, the depletion region is 400nm wide and an electric field of $5 \times 10^5 \text{ V/m}$ exists in it. What should be the minimum kinetic energy of a conduction electron which can diffuse from the n-side to the p-side ?

- 20 eV
 200 eV
 2000 eV
 20000 eV

Key : 3

Solution : $V = E.d$
 $= 5 \times 10^5 \times 400 \times 10^{-9}$
 $= 200 \times 10^{-2}$

$$= 0.2 \text{ volt}$$

$$\therefore \text{Energy} = 0.01 \text{ eV}$$

108. A solid sphere having a coefficient of cubical expansion ‘ α ’ is suspended from a rigid support with a thread in to a beaker containing a liquid of coefficient of real expansion ‘ β ’ such that the sphere is completely submerged in the liquid. On heating the system, the tension in the string is found to be constant. This is possible when

$$\alpha < \beta$$

$$\alpha > \beta$$

$$\alpha = \beta$$

$$\alpha = \beta \times 3$$

Key: 3

$$\text{Solution : } T = mg - \rho_0 V$$

$$\Delta T = \Delta mg - \Delta \rho_0 V$$

Either $\Delta T = 0$ and we know that mg does not change on heating $\Delta mg = 0$

$$\text{So, } \Delta \rho_0 V = 0$$

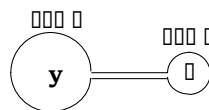
$$\Delta \left(\frac{\rho_0}{d_0} d_0 g \right) = 0$$

$$\Delta \left(\frac{\rho_0}{d_0} d_0 g \right) = 0$$

$$\Delta d_0 = \Delta d_0$$

$$\Rightarrow \alpha = \beta$$

109. In the figure the volume of flask Y is twice that of flask X and is connected by a narrow tube as shown. The system is filled with an ideal gas and the flasks X and Y are kept at 200 K and 400 K respectively. If the mass of the gas in X is “m”, the mass of the gas in Y is



$$\frac{m}{2}$$

$$2m$$

$$m$$

$$\frac{m}{4}$$

Key: 3

$$\text{Solution : } PV = nRT$$

$$\frac{P_X V_X}{P_Y V_Y} = \frac{n_X R T_X}{n_Y R T_Y}$$

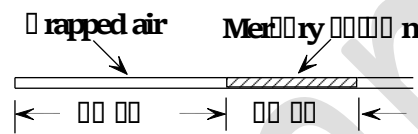
110. Ice, water and steam co-exist at triple point temperature 273.16K and pressure 4.6mm Hg. In a system in which the triple point conditions of temperature and pressure exist, the pressure is increased a little while keeping the temperature constant, then the system contains

- ice only
 water only
 steam only
 water and ice

Key : 2

Solution : At triple point, when pressure is increased, it completely converts into water.

111. A 20 cm column of air is trapped by a column of Hg 15 cm long in a capillary tube of uniform bore when the tube is held horizontally in a room where the atmospheric pressure is 75 cm of mercury. The length of the air column, when the tube is held vertically with open end down will be



- 10 cm
 15 cm
 17.5 cm
 20 cm

Key: 2

Solution : $P_1 l_1 = P_2 l_2$

112. Two spherical black bodies made of same material and with the same surface finish have masses M_1 and M_2 and are at temperatures T_1 and T_2 . If they are radiating the same power, M_1/M_2 must be

- T_1/T_2
 T_1^2/T_2^2
 T_1^3/T_2^3
 T_1^4/T_2^4

Key: 3

Solution : $P = A e \sigma T^4$

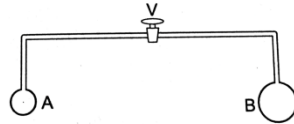
$$A \propto \frac{M}{\rho}$$

$$\Rightarrow (A)^4 \propto \frac{M^4}{\rho^4}$$

$$\Rightarrow (M)^4 \propto \frac{M^4}{\rho^4}$$

$$\Rightarrow \frac{M_1}{M_2} = \left(\frac{\rho_1}{\rho_2} \right)^{\frac{4}{4}} = \left(\frac{\rho_1}{\rho_2} \right)$$

113.



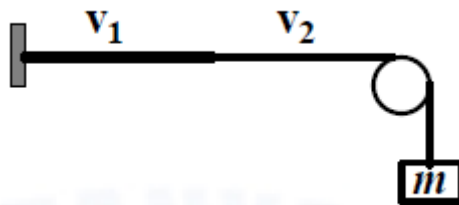
The valve V in the bent tube is initially kept closed. Two soap bubbles A (smaller) and B (larger) are formed at the two open ends of the tube. V is now opened, and air can flow freely between the bubbles.

Which of the following statements is correct?
 (a) The radius of bubble A increases and the radius of bubble B decreases.
 (b) The radius of bubble A decreases and the radius of bubble B increases.
 (c) The radius of bubble A increases and the radius of bubble B increases.
 (d) The radius of bubble A decreases and the radius of bubble B decreases.

Key : 3

Solution : $P \propto \frac{1}{r}$ and
 $P \propto \frac{1}{r}$

114. A weight is hung over a pulley and attached to a string composed of two parts, each made of the same material but one having four times the diameter of the other. The string is plucked so that a pulse moves along it, moving at speed v_1 in the thick part and at speed v_2 in the thin part. What is v_1/v_2 ?

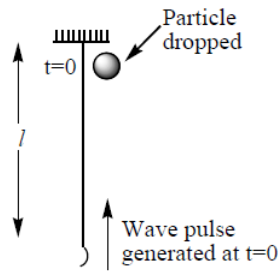


Which of the following is correct?
 (a) $v_1/v_2 = 2$
 (b) $v_1/v_2 = 4$
 (c) $v_1/v_2 = 8$
 (d) $v_1/v_2 = 16$

Key : 3

Solution : $v = \sqrt{\frac{T}{\mu}} \Rightarrow v \propto \frac{1}{\sqrt{\mu}}$

115. A uniform rope of length 's' is suspended from a ceiling as shown. A particle is dropped from the ceiling at the instant a wave pulse is formed at the lower end. Where will the particle meet the pulse?



$\frac{l}{2}$ distance from the top

$\frac{l}{2}$ distance from the bottom

$\frac{l}{3}$ distance from the bottom

at midpoint of the rope

Key : 2

Solution : Let 'x' be the distance from top, (l - x) from the bottom

$$t_{\text{particle}} = t_{\text{wave}}$$

$$\sqrt{\frac{2x}{g}} = \sqrt{\frac{2(l-x)}{g/2}}$$

$$x = \frac{l}{3} \text{ from top}$$

116. A ray of light travels from an optically denser to a rarer medium. Maximum possible deviation is θ . Maximum possible deviation if light travels from rarer to denser is

θ

$\frac{\theta}{2}$

θ

$\frac{\theta}{2}$

Key : 2

Solution : denser to rarer, max. Deviation, $d = (180 - 2C)$

$$\theta = (180 - 2C)$$

$$\theta = 2(90 - C) \rightarrow (1)$$

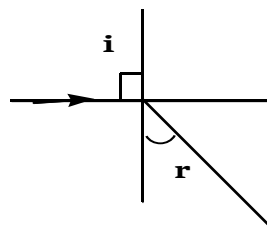
rarer to denser,

$$i = 90^\circ$$

$$r = C$$

$$d = (90 - r) = (90 - C)$$

$$d = \frac{\theta}{2} \rightarrow (2)$$



117. In a Young's double-slit experiment, the central bright fringe can be identified
- (A) as it has greater intensity than the other bright fringes
 - (B) as it is wider than the other bright fringes
 - (C) as it is narrower than the other bright fringes
 - (D) by using white light instead of monochromatic light

Key : D

Solution : Central fringe will only be white in colour

118. A narrow stream of electrons of energy 100eV is fired at two parallel slits very close to each other. The distance between the slits is 10Å. The electron waves after passing through the slits interfere on a screen, 3m away from slits. The fringe width is

- (A) 0.000001 m
- (B) 0.000002 m
- (C) 0.000003 m
- (D) 0.000004 m

Key : 1

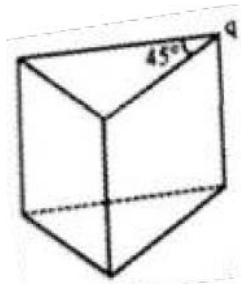
Solution : Wavelength of e^- waves

$$\lambda = \frac{hc}{\sqrt{2mE}} = 0.0000038 \text{ m}$$

Like YDSE, $\beta = \frac{\lambda D}{d}$

$$\beta = \frac{0.0000038 \times 3}{0.01} = 0.000114 \text{ m}$$

119. A prism shaped imaginary structure is given. A point charge 'q' is kept as given in figure. The electric flux passing through the prism is



- (A) $\frac{q}{\epsilon_0}$
- (B) $\frac{q}{2\epsilon_0}$
- (C) $\frac{q}{4\epsilon_0}$
- (D) $\frac{q}{8\epsilon_0}$

Key : 3

Sol : $360^\circ/45^\circ=8$ To enclose the charge one more such set should be placed above it. So total 16 prisms are required.

120. For a closed surface through which the net flux is zero, each of the following four statements *could* be true. Which of the statements *must* be true ?

a) There are net charges inside the surface

b) The net charge inside the surface is zero

c) The electric field is zero everywhere in the surface

d) The number of electric field lines entering the surface equals the number leaving the surface

a) c

b) d

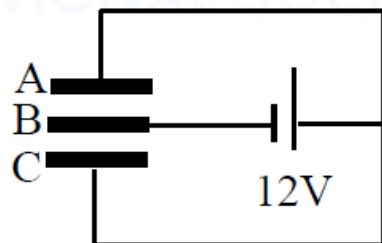
b) c

a) d

Key : 2

Solution : Charges might be there, but net charge is zero inside and $\sigma \neq 0$ on surface.

121. Three plates A,B,C each of area 100cm^2 have separation 3mm between A and B and 3mm between B and C. The energy stored when the plates are fully charged is



$1.2 \times 10^{-6}\text{ J}$

$1.2 \times 10^{-5}\text{ J}$

$1.2 \times 10^{-4}\text{ J}$

$1.2 \times 10^{-3}\text{ J}$

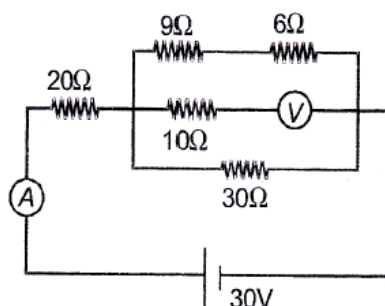
Key : 2

Solution : 2 Capacitors in parallel

$$U = \frac{1}{2} (nC) V^2$$

$$C = \left(\frac{\epsilon_0 \epsilon_r A}{d} \right) V^2$$

122. In the circuit shown in figure, if ammeter and voltmeter are ideal, then the power consumed in 10Ω resistor will be



1.2 W

1.2 W

1.2 W

1.2 W

Key : 2

Solution : If volt meter ideal

$R = \infty$, so no current through volt meter branch.

$$R_{\text{eq}} = 2R$$

$$i = \frac{E}{2R} = \frac{E}{2R}$$

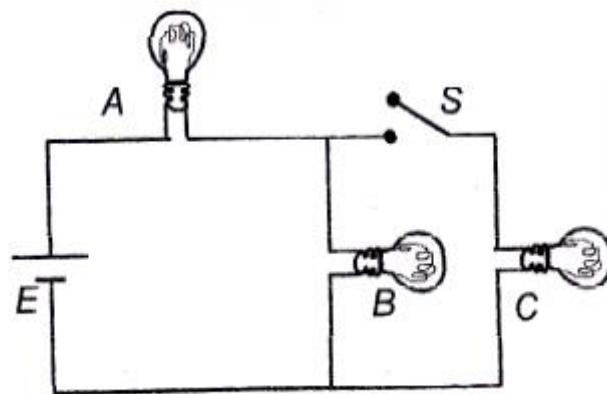
$$\text{In } R \text{ branch } i_1 = \left(\frac{E}{2R}\right) \times \frac{1}{2}$$

$$i_1 = \frac{E}{4R}$$

Power is 'P', $P = i_1^2 R$

$$P = \left(\frac{E}{4R}\right)^2 \times R = \frac{E^2}{16R}$$

123. Three identical bulbs are connected as shown in figure. When switch S is closed, the power consumed in bulb B is P. What will be the power consumed by the same bulb when switch S is opened ?



$$\frac{P}{4}$$

$$\frac{P}{2}$$

$$P$$

$$2P$$

Key : 1

Solution : When switch is closed

$$\mathcal{E} = \frac{\Delta \Phi}{\Delta t}$$

$$\text{Total current, } i = \frac{\mathcal{E}}{R}$$

$$\text{Current in bulb 'B', } i_b = \frac{i}{2}$$

$$i_b = \frac{\mathcal{E}}{2R}$$

$$\text{Power of bulb 'B', } P = i_b^2 R$$

$$P = \frac{\mathcal{E}^2}{4R} \rightarrow (\text{D})$$

When switch 'S' opened

$$\mathcal{E} = 0$$

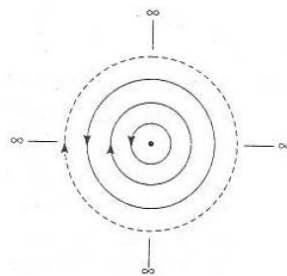
$$\text{Total current, } i = 0$$

$$\text{Power in 'B', } P' = i^2 R$$

$$P' = 0 \rightarrow (\text{C})$$

$$\text{ss } \frac{P}{P'} = \frac{\mathcal{E}^2}{0} \Rightarrow \boxed{P' = \frac{P}{4}}$$

Four infinite conducting rings each having current I in the directions shown are placed concentrically in the same plane as shown in the figure (currents in successive coils are in opposite directions of rings). The radii are $r, 2r, 3r, 4r$. The total magnetic induction at their common centre is



$$\frac{\mu_0 I}{2r}$$

$$\frac{\mu_0 I}{r}$$

$$\frac{\mu_0 I}{2r}$$

$$\frac{\mu_0 I}{r}$$

Key : 4

Solution $B_C = \frac{\mu_0 I}{2r}$ due to each circular loop

125. A charged particle moves undeflected in a region of crossed electric and magnetic fields. If the electric field is switched off, the particle has an initial acceleration a . If the magnetic field is switched off, instead of the electric field, the particle will have an initial acceleration

- a
 a
 a
 a

Key : 3

Solution: Initial velocities are same

126. Into a transverse uniform magnetic field of induction 6.5 G an electron is projected with a speed $\times 10^6$ m/s, at angle $^\circ$ to the boundary. The time elapsed by the electron in the field in nano second is

- infinity

Key : 3

Solution : It comes out of field after it completes half revolution

$$t = \frac{\pi r}{v} = \frac{\pi \left(\frac{mv}{eB} \right)}{v}$$

$$t = \frac{\pi m}{eB}$$

127. In LCR circuit current resonant frequency is 600Hz and half power points are at 650 Hz and 550Hz. The quality factor is

-

Key : 3

Solution : $Q = \frac{\omega}{\Delta\omega}$

$$= \frac{600}{650 - 550} = 6$$

128. A capacitor of capacitance C is given charge Q and then connected in parallel to a coil of inductance L . There is no resistance in the circuit. When the charge on the capacitor becomes zero, the current in the coil will be

- $Q\sqrt{\frac{L}{C}}$
 $\frac{Q}{\sqrt{LC}}$
 $Q\sqrt{\frac{C}{L}}$
 erf

Key : 2

Solution : $\frac{Q^2}{2C} = \frac{1}{2} LI^2$ y conservation of energy

129. In a uniform magnetic field of T in free space, the energy density is u . The electric field which will produce the same energy density in free space is

$$\frac{1}{\epsilon_0} \int \rho \, dV$$

$$\frac{1}{\epsilon_0} \int \rho \, dV$$

$$\frac{1}{\epsilon_0} \int \rho \, dV$$

$$\frac{1}{\epsilon_0} \int \rho \, dV$$

Key : 2

Solution: $U = \frac{1}{2} \epsilon_0 \int E^2 \, dV$ and $C = \frac{1}{\sqrt{\epsilon_0 \mu_0}}$

130. A bar magnet is demagnetized by inserting it inside a solenoid of length 0.2m, 100 turns, and carrying a current of 5.2 A. The coercivity of the bar magnet is:

$$1000 \text{ A/m}$$

$$10000 \text{ A/m}$$

$$100000 \text{ A/m}$$

$$1000000 \text{ A/m}$$

Key : 2

Solution: $B = \mu_0 nI$

$$H = nI$$

131. An example of a perfect diamagnet is a superconductor. This implies that when a superconductor is put in a magnetic field of induction B, the magnetic field B_s inside the superconductor will be such that:

$$B_s = -B$$

$$B_s = 0$$

$$B_s = B$$

$$B_s < B \text{ and } B_s \neq 0$$

Key : 2

Solution ; For super conductors, $\chi = -1$ Since $\mu_r = 1 + \chi \Rightarrow \mu_r = 0$

A small ball is projected with initial speed u and at an angle θ with horizontal from ground. The de – Broglie wave length of ball at the moment its velocity vector becomes perpendicular to initial velocity vector is

$$\frac{h}{mu}$$

$$\frac{h}{mu \sin \theta}$$

$$\left(\frac{h}{mu} \right) \tan \theta$$

$$\frac{h}{mu \cos \theta}$$

Key : 3

Solution: The velocity of projectile when its initial velocity is perpendicular to final vector is

$v = u \cot \theta$ de-Broglie wave length $\lambda = \frac{h}{mv}$

$$\lambda = \frac{h}{mu \cot \theta}$$

$$\lambda = \left(\frac{h}{mu} \right) \tan \theta$$

133. An energy of 24.6 eV is required to remove one of the electrons from a neutral helium atom. The energy (in eV) required to remove both the electrons from a neutral helium atom is

$$24.6 \text{ eV}$$

$$49.2 \text{ eV}$$

$$73.8 \text{ eV}$$

$$98.4 \text{ eV}$$

Key : 4

Solution : $\lambda = \frac{h}{mv} \times \frac{v_0}{v_0}$

Z = 2

134. An electron (mass m) with an initial velocity $\mathbf{v} = v_0 \hat{i}$ is in an electric field $\mathbf{E} = E_0 \hat{j}$. If initial wavelength $\lambda_0 = \frac{h}{mv_0}$, then its de Broglie wavelength at time t is given by

λ_0
 $\lambda_0 \sqrt{1 + \frac{eE_0 t}{mv_0}}$
 $\frac{\lambda_0}{\sqrt{1 + \frac{eE_0 t}{mv_0}}}$
 $\frac{\lambda_0}{\sqrt{1 + \frac{eEt}{mv_0}}}$

Key : 3

Solution : $V = u + at$

$= V_0 \hat{i} - \frac{e}{m} E_0 t \hat{j}$

$V = \sqrt{V_0^2 + \frac{e^2 E_0^2 t^2}{m^2}}$

$\lambda = \frac{h}{mv}$

$= \frac{h}{m \sqrt{V_0^2 + \frac{e^2 E_0^2 t^2}{m^2}}}$

$= \frac{h}{mv_0 \sqrt{1 + \frac{e^2 E_0^2 t^2}{m^2 V_0^2}}}$

$= \frac{\lambda_0}{\sqrt{1 + \frac{e^2 E_0^2 t^2}{m^2 V_0^2}}}$

135. At $t = 0$, a radioactive substance has a mass m . Its half-life is 10 minutes. When $t = t_1$, the amount of substance disintegrated is $m/5$ and when $t = t_2$, the amount of substance disintegrated is $3m/5$. Then the time interval $(t_2 - t_1)$ is

- $10 \ln 5$ in tes
 $10 \ln 2$ in tes
 $10 \ln 3$ in tes
 $10 \ln 15$ in tes

Key : 1

Solution : At t_1 remaining mass $\frac{M}{2}$

At t_2 remaining mass $\frac{M}{4}$

In time interval $(t_2 - t_1)$ it is reduced to half.

$(t_2 - t_1) = \tau = 10$ in.

...



SRI CHAITANYA EDUCATIONAL INSTITUTIONS,INDIA.

A.P,TELANGANA,KARNATAKA,TAMILNADU,MAHARASHTRA,DELHI,RANCHI,CHANDIGARH

SEC : SR ELITE (SET-1)

DATE : . .20

SUB : CHEMISTRY

NEET GRAND TEST-7

Identify the correct statements of the following

1. Oxygen has less electronegativity

2. Lanthanides and actinides have largest atomic radii

3. The strongest oxidizing agent is reduced by hydrogen

4. Among Si, Mg and P the most electropositive element is Mg

5.

6.

7.

8.

9.

10. Among Si, Mg and P the most electropositive element is

11. Positive electron gain enthalpy is highest for

12. e

13. r

14. e

15. r

16.

17. Among the noble gases the gas element is e r r e n e

18. Piezoelectric standard the following is

19. urea

20. quartz

21. graphite

22. Charcoal

23.

24. Among quartz is used as piezoelectric standard

25. Among the following at this

26. Mn antiferromagnetic

27. all substances

28. Mg e_g , ferromagnetic

29. Cr₂O₃ , ferromagnetic

30.

31. Among the following is considered as Se indicator

32. For high temperature the following reaction the rate constant increases with temperature

33. element present in the catalyst is dioxin in a chemical reaction

34. reaction rate increases with temperature and

35. reaction rate increases with temperature and

36. element present in the gas phase is platinum at high pressure

37.

Stoichiometric position of Pt in PtCl₄ is $\frac{1}{4}$ in PtCl₄ at high pressure is $\frac{1}{4}$ in order reaction the units are

order is $\frac{1}{4}$ the rate is $\frac{1}{4}$

high the reaction is irreversible

S₀ is oxidizing while S₂ is reducing agent

In vapour state S₂ molecule is para magnetic $\uparrow \uparrow$

has S₀ - S₂ linkage

key step in that process and rate is $S_{(g)} + S_{(g)} \rightleftharpoons S_{2(g)}$

Key

Stoichiometric of acts as oxidizing agent

ring at this

Cr - Melting point

Cr(+3) > Mn(+3) > Fe(+3) stability oxidation state

Cr³⁺ < Cr²⁺ < Mn²⁺ oxidizing power

Cr - Mn electrode potential (M³⁺ / M²⁺)

Key

Stoichiometric the order stability oxidation state is Cr > Mn > Fe

the p-bonds $\begin{array}{c} \parallel \\ \text{O} \\ | \\ \text{C} - \text{O} - \text{C} - \text{O} \end{array}$ and $\begin{array}{c} \parallel \\ \text{O} \\ | \\ \text{C} - \text{O} - \text{C} \end{array}$ are

terminal isomers

all isomers

Meta isomers

Chain isomers

Key

Stoichiometric $\begin{array}{c} \parallel \\ \text{O} \\ | \\ \text{C} - \text{O} - \text{C} - \text{O} \end{array}$ and $\begin{array}{c} \parallel \\ \text{O} \\ | \\ \text{C} - \text{O} - \text{C} \end{array}$ are eta isomers

graphs are eta number as they are completely displaced by C the equivalent eight eta is

terminal

terminal

terminal

terminal

Key

$$\text{Stoichiometric} \frac{\text{terminal}}{\text{terminal}} = \frac{\text{terminal eta number}}{\text{terminal eta number}}$$

$$\frac{K_{sp} \times K_{a2}}{K_{a1}} = \frac{K_{a2}}{K_{a1}}$$

∴ $\frac{K_{a2}}{K_{a1}}$ is the K_{sp} of H_2PO_4^-
 $\frac{K_{a2}}{K_{a1}}$ is the K_{sp} of H_2PO_4^-
 $\frac{K_{a2}}{K_{a1}}$ is the K_{sp} of H_2PO_4^-
 $\frac{K_{a2}}{K_{a1}}$ is the K_{sp} of H_2PO_4^- – Ans 39

For the solubility of CaF_2 in H_2O is given by

$\text{CaF}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{F}^-$

$\text{CaF}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{F}^-$

$\text{CaF}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{F}^-$

$\text{CaF}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{F}^-$ and $\text{CaF}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{F}^-$

Ans

For the solubility of CaF_2 in H_2O is given by $\text{CaF}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{F}^-$ and $\text{CaF}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{F}^-$

For a galvanic cell the cell notation is $\text{Zn}(s) | \text{Zn}^{2+}(aq) || \text{Cu}^{2+}(aq) | \text{Cu}(s)$ and the highest

of the values and show the

Zn^{2+}	Cu^{2+}
1 M	1 M
1 M	1 M
1 M	1 M
1 M	1 M

Ans

For the cell $\text{Zn}(s) + \text{Cu}^{2+}(aq) \rightarrow \text{Zn}^{2+}(aq) + \text{Cu}(s)$

$$E = E^\circ - \frac{RT}{n} \ln \frac{[\text{Products}]}{[\text{Reactants}]}$$

$$= E^\circ - \frac{RT}{n} \ln \left[\frac{[\text{Zn}^{2+}]}{[\text{Cu}^{2+}]} \right]$$

$$E = E^\circ - \frac{RT}{n} \ln y$$

$$\therefore \alpha = \frac{E}{RT \ln y}$$

For the solubility of CaF_2 in H_2O is given by $\text{CaF}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{F}^-$ and $\text{CaF}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{F}^-$

1 M 1 M $1 \times 10^{-4} \text{ M}$ 1 M

Ans

For the cell $\text{Zn}(s) | \text{Zn}^{2+}(aq) || \text{Cu}^{2+}(aq) | \text{Cu}(s)$

$$E \times 10^{-4} = S(s + \text{RT})$$

$$S \times 10^{-3} = S^0 + S \times 10^{-3}$$

$$S \times 10^{-3} = S \times 10^{-3}$$

$$S = S \times 10^{-3} M$$

the following are antitubercular antibiotics

Penicillin

Isoniazid

Streptomycin

Rifampin

Q. 10

State the function of ethylamine

Phenylamine, Picramide and neoprene are all

condensation polymers

addition polymers

biodegradable polymers

addition polymers

Q. 11

State the function of Picramide and neoprene are addition polymers

Highly the following vitamins are antioxidants

Vitamin A

Vitamin B

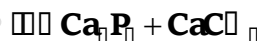
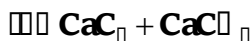
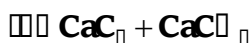
Vitamin C

Vitamin D

Q. 12

State the function of Vitamin A

Holme's signals can be obtained by using



Q. 13

State the function of $CaC_2 + Ca_3P_2$ used in the Holme's signals.

Salt with high negative charge is precipitated with potassium

hydroxide



with and

Q. 14

State the function of P^{+3} and a^{+2} ions give yellow ppt with potassium hydroxide

In the refining of ethyl alcohol are reduced in trap reformed he element and the

Si

e

a

e

Q. 15

State the function of the elements Si and a are purified by the refining of ethyl

Magnesium is nitrate in drinking water is ppm and its excess concentration causes

ppm kidney damage

ppm electrolyte syndrome

pp t th delay

pp a active e l t

ey

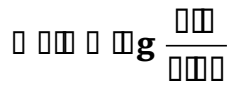
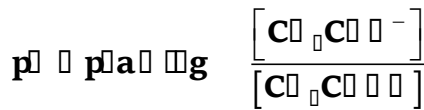
S t n M a i i i t i n d r i n g a t e r i s p i t a s e a y s y n d r e

n e l i t e a n a e s s t i n n t a i n e s C C p a and e

C C a h e p s t i n i s

ey

S t n



e p s i t i n is p r e v e n t e d y

M n

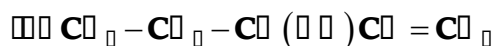
e t a n i d e

P t

ey

S t n e g a t i v e a t a l y s t p r e v e n t s d e p s i t i n e t a n i d e a t s a s n e g a t i v e a t a l y s t r d e p s i t i n .

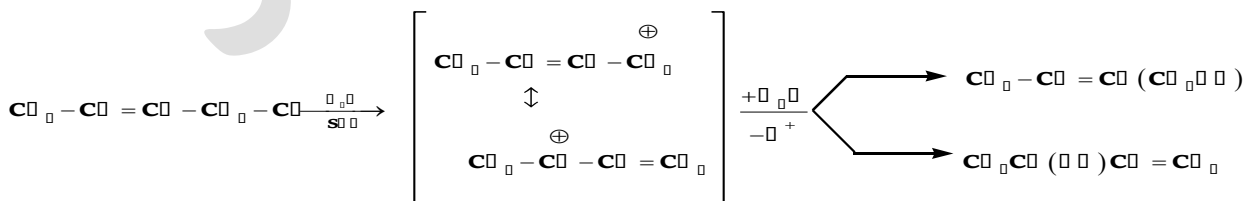
C C - C C = C C - C C - C C x is e a l t i n p r e e d s y S e h a n i s



th and

ey

S t n



h i h n e t h e i n g n h y d r o l y s i s g i v e s t h e r e s p o n d i n g e t a n i h y d r i d e a n d

i

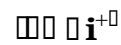
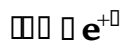
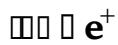
a

e

Q. 10

Statement: $e^- + e^- \rightarrow e^- + e^- + e^-$

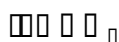
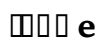
Neils Bohr's atomic model is not applicable to



Q. 11

Statement: Bohr's atomic model is applicable to Hydrogen and hydrogen like ions

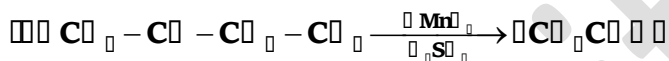
Value of λ in Balmer's series is in



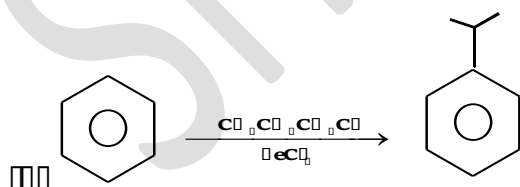
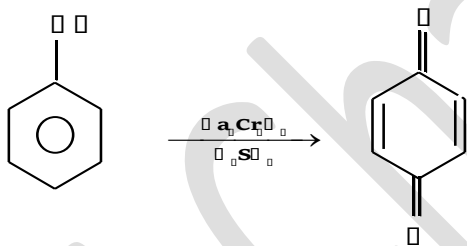
Q. 12

Statement: C_6H_6 has least λ in Balmer's series if it has least critical temperature

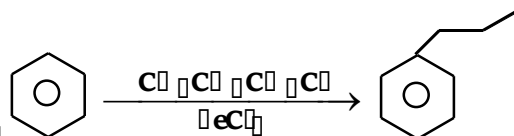
High λ is not a correctly represented reaction



Q. 13

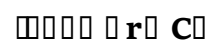
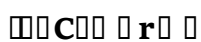


Q. 14



Statement:

Order of increasing order of stability of oxides is



Q. 10

Stability order of the following is

the following pair of species in high base Δ_0 value is higher for the second one



Q. 11

Stability order of Δ_0 is

$Cr^{3+} > Cr^{2+} > Cr^{+}$

High the following is the best electron releasing group when present in enone ring



Q. 12

Stability order of $-C_2H_5$ has α hydrogens. It has more hyperconjugative

groups activated hydrogens as added to the acetone molecule in a gas phase than in a liquid phase. The strength of the C-H bond as a function of the amount of acetone added per gram of hydrocarbon is



Q. 13

Stability order of the standard enthalpy of



\therefore 1000 g

Stability order in ethyl carbonyls

increases para magnetism

increases M-C bond length

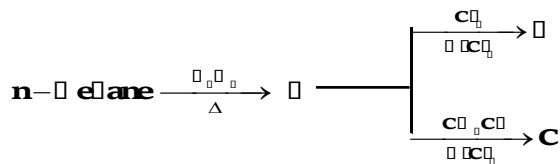
increases C-C bond length

increases M-C bond order

Q. 14

Stability order of Syngas in ethyl carbonyl increases M-C bond order $M-C \equiv C \Rightarrow M-C \equiv C$

Q. 15

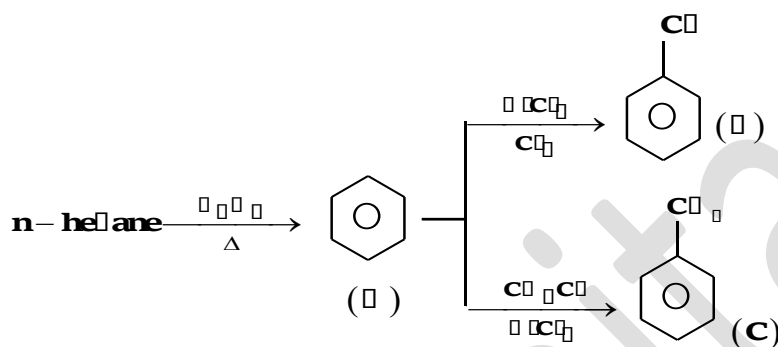


Order of reactivity towards electrophilic substitution

Cyclohexane
Benzene
Cyclohexene
Cyclohexadiene
Cyclohexane

Key

Substitution



Order of reactivity C > C=C > C=C=C

State entropy of fusion $\Delta S^\circ_{\text{fus}}$ always less

State entropy of fusion at STP $\Delta S^\circ_{\text{fus}}$ of H_2O and H_2O are equal

Both State entropy and State entropy are more

State entropy is more State entropy is more

Both State entropy and State entropy are more

State entropy is more State entropy is more

Key

Substitution of fusion $\Delta S = 0$ and $\Delta S^\circ \neq 0$

$$\text{Molar heat capacity at STP } \frac{n}{v} = \frac{3}{2} R$$

$$\text{Molar heat capacity at STP } \frac{n}{v} = \frac{5}{2} R = \frac{5}{2} \times 8.314 \text{ J K}^{-1} \text{ mol}^{-1}$$

Heat capacity of C_6H_6 , C_6H_6 , C_6H_6 , C_6H_6 , C_6H_6 , C_6H_6 are more or less

Order of reactivity

Cyclohexane
Benzene
Cyclohexene
Cyclohexadiene

Key

Solution C_0 and C_1 undergo Cannizzaro's reaction

When gram substance having the empirical formula C_0 is dissolved in g water the

solution freezes at $0^\circ C$ that is the molecular formula $= 100^\circ C/g$



Key

Solution $\Delta T_f = 0$

$$\Delta T_f = \Delta T_f \times \frac{100}{M} \times \frac{1}{100}$$

$\therefore M = 100$

$M = n \times 100$

$$n = \frac{M}{\text{empirical formula weight}}$$

$n \times (C_0)$

$= \frac{100}{100} = 1$

$n \times (C_1)$

C_1

It reacts with Hinsberg's reagent and gives a precipitate in a definite amount

amide hydrogen is

iline

ethyl aniline

Methylaniline

idine

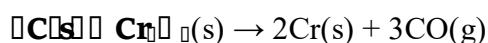
Key

Solution Amines react with Hinsberg's reagent to give sulphanamide which is insoluble in NaOH

The enthalpy changes for the reactions are given by the equations



That is the enthalpy change in the reaction



1000000

1000000

1000000

1000000

Ans

Solution Heat change of a reaction = $\Delta H = (\text{Sum of the Heats of formation of products}) - (\text{the$

reactants)

$$\Delta H = [2(0) + 3(100000 - 100000)]$$

The highest oxidation state exhibited by an element is

Strong acid

Weak acid

Moderately strong acid

Triethyl acid

Key

Strong acids having pKa value between 0 are moderately strong acids

High the pKa is a general protein

seratin

Insulin

Myosin

th and

Key

Strong insulin is a general protein

13th group element with least melting point is

a

h

Key

Strong Ga has least melting point among 13th group elements.

1 mole of propene on treatment with 'x' moles of O_2 is oxidized by treatment with C_2O_2 r

gives a pentyne. The value of 'x' is

Key

Strong

